

Antioxidant and Trace Metals Levels in Anemia

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ABSTRACT

Anemia is a disorder characterized by reduction in hemoglobin concentration of blood below the normal level. It is a global problem, mainly affecting poor people in developing countries. For the determination of these metals fifty intravenous blood samples anemic patients and normal subjects were collected and immediately centrifuged to obtain the supernatant liquid, serum of both the groups for analysis. The mean serum levels of magnesium, zinc, copper, and iron and antioxidant activity in anemic patients were 16.05 ± 1.14 ppm, 4.33 ± 0.28 ppm, 2.5 ± 0.2 ppm, 1.50 ± 0.28 ppm and 1.27 ± 0.05 ppm respectively. Serum magnesium, zinc, iron and antioxidant levels were significantly lower whereas, the higher level of Copper was observed in anemic patients as compared to the controls. Supplementation with deficient trace elements may reduce the severity of symptoms and complications associated with anemia.

Key Words

Anemia, Iron, Zinc, Copper, Magnesium, Antioxidant activity.

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1 Introduction

Anemia is one of the world's most prevalent health problems, especially among adult and pregnant females approximately 35% to 75% (56% on average) are anemic in developing countries (Hall A and Miguel E (2001). "The World Health Organization has identified that iron and folic acid deficiency is a severe problem in Pakistan and it has devastating impacts on health and survival. It has been estimated that iron deficiency leads to 22,000 maternal deaths for the last 10 years (ADB 2006). Whereas there are various other nutritional and communicable causes of anemia, iron (Fe) deficiency is often a causative aspect in many of these cases (Lanzkowsky P 1995). Fe deficiency anemia leads to weakness (Haas JD, 2001), poor physical growth (Grantham-McGregor S, 2001), and delayed psychomotor development (Sherriff A, 2001). A recent study on the prevalence and etiology of nutritional anemia in urban areas of Hyderabad, Pakistan indicated a high prevalence of anemia and Fe deficiency in 60% of adults. Zinc (Zn) deficiency in humans has been found in infants, schoolchildren and adults (Castillo-Duran C, 1994). Zinc deficiency associates with Fe deficiency anemia in areas where Fe deficiency is a problem (Singh S, 2003). It was investigated that anemia is one of the major manifestations of copper (Cu)

deficiency both in animals and human beings (Percival SS, 1995). It is essential to maintain Fe homeostasis in the human body; its deficiency leads to anemia and neutropenia (Tammura H, 1994). Cu-assisted enzymes are necessary for the utilization of Fe to make hemoglobin, a main component of red blood cells (Tammura H, 1994). Recent attention has been directed to the element chromium (Cr); it can improve insulin sensitivity and therefore may be involved in carbohydrate and lipid metabolism (Lukaski HC, 1999). The determination of trace quantities of metals in biological samples requires the use of sensitive and selective techniques such as atomic absorption spectrometry.

In this study, we evaluate whether Fe, Cu, Mg and Zn deficiency is related to increased level of toxic metals in blood samples of anemic and controls of both gender with age range 16–50 years. The samples were prepared method. Fe, Cu, Mg and Zn concentrations in samples under study were evaluated by atomic absorption spectrometer, while antioxidant activity was analyzed by spectrophotometer.

In this study, we aimed to evaluate the levels of Cu, Mg, Zn, and Fe and antioxidant activity in serum of anemic patients.

2 Materials and Methods

This investigation has been

conducted among three medical units of Liaquat University of Medical & Health Sciences (LUMHS) Jamshoro. The group of patients were been selected within the age range of 16-50 years of both gender. The metals copper, iron, magnesium, and zinc in the blood serum were determined by Atomic Absorption Spectrometry (AAS) (Model, A-20 Varian). Whereas anti oxidative activity was measured as a ratio of Fe (II)/Fe (III) present in the blood serum. Iron (II) was determined spectrophotometrically as Fe (II)-TPTZ (2, 4, 6-tris (2-pyridyl)-s-triazine) colored complex formed at pH 4-5 and measured at 595nm wavelength. Whereas, for the determination of total iron, iron (III) was first reduced to Fe⁺⁺ using vitamin C as reducing agent and then complexed with TPTZ. The complex so produces is directly proportional to the concentration of Fe (II) produced and Fe (II) already present i.e. total iron in the blood sample in presence of antioxidant. The difference of the total iron and the iron (II) present in the serum is the measure of iron (III).

Trace metals were determined using airacetylene flame. The standards from 1 to 5 ppm for each of the metal separately were run on the spectrometer and the calibration curves were obtained prior to run the samples for the determination of metals in the blood serum of normal subjects and the malarial patients. Blood samples were collected from 50 healthy controls in fasting conditions and a similar condition was maintained while taking blood samples of anemic patients. Each blood sample was centrifuged at 5000 rpm for 20 minutes. The supernatant blood serum was used for the analysis of metals copper, iron, magnesium, and zinc using Atomic Absorption Spectrometer inserting appropriate hollow cathode lamp in it. All standards used were of analytical grade.

Chemicals and reagents

Sulphosalicylic acid was obtained from Merck, Damstadt, Germany and other chemicals to prepare standards were purchased from Sigma Chemical Co. All chemicals were of analytical

grade.

Stock Solutions and working Metal standards

Stock solution of 1000 ppm Cu, Fe, Mg, and Zn for each were prepared for corresponding sulphate salts of analytical grade (Sigma Chem.). Working standards were prepared from the stock solutions by diluting with appropriate volume of deionized water and addition of few drops of corresponding concentrated acid.

Statistical Analysis

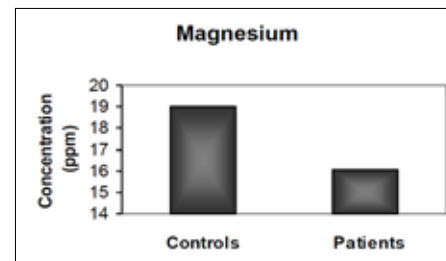
All statistical analyses were performed using computer program Excel X State (Microsoft Corp., Redmond, WA) Student's t test was used to assess the significance. Results were expressed as mean \pm SD.

3 Results

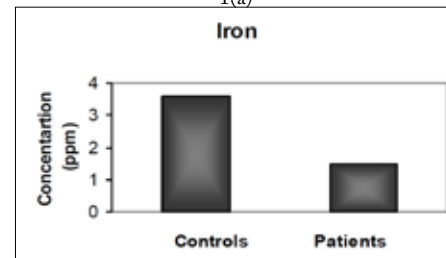
Results and Discussion

Table 1 shows the blood serum levels of trace metal and antioxidant levels in anemic patients with compare to control subjects. The results show significant increase in serum copper levels, whereas serum iron, magnesium, zinc and antioxidant levels were decreased in anemic patients as compared to the controls.

The Table 1 show the levels of trace metals (Fe, Cu, Mg and Zn) and antioxidant activity were evaluated in blood samples of anemic patients as compared to controls. Fe deficiency is probably the most common nutritional disorder in the world. Presently estimate based on the WHO criteria specify that around 600-700 million people worldwide have a marked Fe deficiency anemia and about half the adults in developing countries are affected (Oski FA, 1993). The data of the trace metals under study in blood samples of control and anemic patients are shown in Table 1 and Figure 1(a) to 3(a). The results are given as mean values with standard deviation (\pm SD)

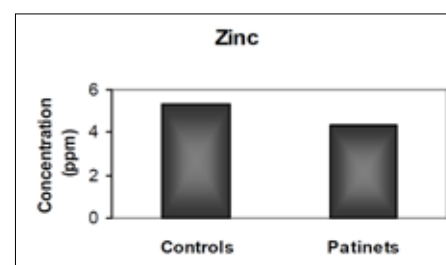


1(a)

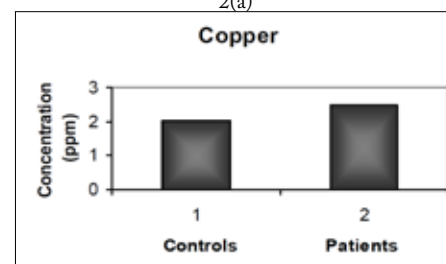


1(b)

Fig.1. 1(a) the decreased level of Magnesium in serum of anemic patients whereas figure 1(b) the decreased level of Iron in serum of anemic patients as compared to control subjects.

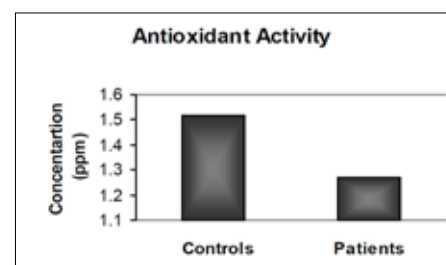


2(a)



2(b)

Fig.2. 2(a) The decreased zinc level in serum of anemic patients and in figure 2(b) the elevated level of copper in serum of anemic patients as compared to control subjects.



3(a)

Fig.3. 3(a) the decreased antioxidant level in serum of anemic patients as compared to control subjects.

Table 1. Trace metals and antioxidant levels in Healthy Controls and Anemic.

Variables	Controls	Anemic
Mg	19.01 \pm 0.68	16.05 \pm 1.14
Iron	3.60 \pm 0.23	1.50 \pm 0.28
Zinc	5.34 \pm 0.39	4.33 \pm 0.28
Copper	2.01 \pm 0.11	2.5 \pm 0.2
Anti Oxidant Activity	1.52 \pm 0.05	1.27 \pm 0.05

for each metal and antioxidant activity. The mean values of iron in the anemic patients were significantly lower as compared to healthy controls (Fig: 1(b)). The results indicate the prevalence of Fe deficiency anemia in both genders. In the population of developing countries, the amount of Fe absorbed from the diet is not sufficient to meet many individuals' requirements. The consumption of a predominantly cereal-based diet, rich in phytate, oxalate, phosphate, fiber, and other inhibitors of Fe absorption, was the main cause of Fe deficiency diseases. Phytates strongly inhibit Fe absorption in a dose-dependent fashion, and even small amounts of phytates have a marked effect (Hallberg L, 1989). If the amount of absorbable Fe in the diet cannot be immediately improved, Fe supplement must be included in the diet to control Fe deficiency anemia (Lozoff B, 1991). The concentrations of Zn in anemic patients were found to be significantly lower as compared to healthy controls (Fig: 2(a)). Zn deficiency may be a contributing factor in anemia (Sondstrom B, 1990). Zn deficiency in developing countries is due to low consumption of meat and fish along with food rich in phytate. Food rich in phytate significantly reduce the absorption of Zn, increasing the chance of Zn deficiency. The higher level of Cu was observed in anemic patients as compared to the healthy controls (Fig: 2(b)). Copper is required for normal infant development, red and white blood cell maturation, Fe transport, bone strength, cholesterol metabolism, myocardial contractility, glucose metabolism, brain development, and immune function (L'Abbe MR, 1992). A deficiency of either Fe or Cu will result in anemia, namely, Fe deficiency anemia or Cu deficiency anemia. Copper is essential for the functioning of many Cu-dependent enzymes (Larsson S, 1995) such as ceruloplasmin (responsible for antioxidant protection, Fe metabolism, and Cu transport), and it was established that the anemia appears to be related to defects in Fe mobilization due to the combined defect of both red ceruloplasmin ferroxidase activity and intracellular utilization (Tapiero

H, 2003). Cu is a major component of catalytic centers of different redox enzymes, and thus, its presence is essential for normal physiologic function such as cellular respiration, free radical defense, synthesis of melanin pigment, connective tissue biosynthesis, and cellular Fe metabolism (Gacheru N, 1990). About 95% of the Cu in the blood is bound to ceruloplasmin. These enzymes play a role in the regulation of Fe metabolism.

The mean values of magnesium in the anemic patients were significantly lower as compared to healthy controls (Fig: 1(a)). Low serum magnesium causes potassium, calcium and neuromuscular disturbances, central nervous system and cardiovascular alterations, like arrhythmias (Swaminathan R, 2003). Furthermore, it can alter glucose homeostasis, increase atherosclerosis, hypertension, myocardial infarction, osteoporosis, migraine, asthma, chronic fatigue syndrome, among others (Swaminathan R, 2003) and Barbagallo M, 2007). More research on the effects of magnesium deficiency on the health of people is needed to warrant interventions to prevent it. The mean values of Anti-oxidant activity in the anemic patients were significantly lower as compared to healthy controls (Fig: 3(a)). Anti-oxidant activity which shows decreasing tendency in patients and this parameter could be used as a biomarker for obtains the status of these patients under medical therapy and its affectivity.

Conclusion

It suggests that use of magnesium as a supplementary diet and use antioxidant activity as the status of drug response to greater number of patients. The decreased levels of iron and zinc can be maintained by given supplement of these metals as therapy.

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